SECONDARY STRUCTURE OF SEA ANEMONE TOXINS

Circular dichroism, infrared spectroscopy and Chou-Fasman calculations

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1. Introduction

Characteristic features of polypeptide toxins isolated from various species of sea anemone are low M_r , high content of hydrophobic amino acid residues, strong basicity and occurrence of 3 disulfide linkages in the molecule [1]. As physiological specificity of toxins may be related to their conformational states different methods were used to study spatial arrangements of toxin II from Anemonia sulcata (ATX-II) and anthopleurin A (AP-A) from Anthopleura xanthogrammica [2-5]. Preliminary data on the conformational stability of toxin I from Radianthus macrodactylus* (RTX-I) appeared in [6]. Here a detailed investigation of circular dichroism (CD) spectra and infrared (IR) spectrum of RTX-I has been performed and an attempt made to generalize the conformational data obtained by spectral methods [2-6] and modified Chou-Fasman method [7] for all toxins considered in [1].

2. Experimental

RTX-I was isolated as in [8]. The carboxymethylated RTX-I was obtained as in [9]. CD spectra were recorded on a Dichrograph III (Jobin-Yvon) in the region of 185-250 nm in aqueous solutions at 1 mg toxin/ml. IR spectra were recorded on a Perkin-Elmer 180 spectrophotometer in matched CaF_2 window cells of 0.078 mm thickness at 10 mg RTX-I/ml in 0.1 M NaCl solution in D_2O . For toxins with the known amino acid sequences [1] the modified Chou-Fasman method [7] was used with Levitt conformational

parameters built up on an extended basis of the data [10]. To predict β -turns we have calculated the product $N_t = P_{t1} \cdot P_{t2} \cdot P_{t3} \cdot P_{t4}$ making use of the Chou-Fasman data [11].

3. Results

3.1. Circular dichroism

CD spectra of proteins in the peptide region (185-250) are directly related to their secondary structure [12]. CD spectrum of RTX-I aqueous solution (fig.1) has a positive band (186 nm) and a negative band (201 nm) of nearly equal amplitudes both due to the $\pi - \pi^*$ amide chromophore transitions. The $n-\pi^*$ peptide transition in the RTX-I CD spectrum is observed as a shoulder at ~ 217 nm. A weak positive CD band at 232 nm may be assigned to transitions of tyrosine and/or tryptopan residues by analogy with CD spectra of snake toxins [13], CD spectrum of the carboxymethylated toxin I (cmRTX-I) is also shown in fig.1. It has a negative maximum at 196 nm and a shoulder at ~220 nm. The absence of the 232 nm band is indicative of a change in the environment of aromatic amino acid residues on reduction of the disulfide linkages in the native RTX-I. The comparison of the obtained CD spectrum of RTX-I with that of AP-A [3] brings out only slight differences in wavelengths and intensities of the CD maxima which confirms once again the assumption of a close relationship between the spatial structure of RTX-I and the other sea anemone toxins [6].

As 'infinite' polypeptide models were used to analyse the secondary structure of proteins from CD spectra [12], some difficulties arise when attempting to determine small segments of the regular structures of short-chain proteins. In [14] model spectra of the

^{*} This sea anemone was defined in [6] as Homostichantus duerdemi, but more accurate analysis has shown it to be Radianthus macrodactylus

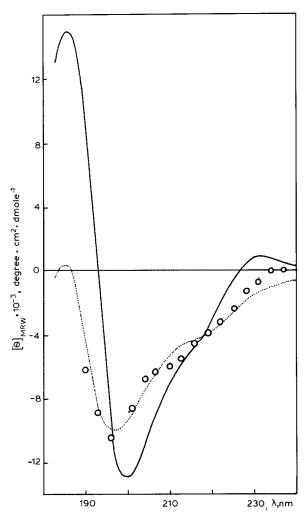


Fig.1. CD spectra of native (——) and carboxymethylated (...) RTX-I in aqueous solutions. (o) Theoretical spectrum of sea anemone toxin calculated on the basis of model CD spectra [14] and data of modified Chou-Fasman method [7].

 α -helices had a limited number of peptide units. Making use of those data and the results obtained by the Chou-Fasman method (table 1) we have calculated the CD spectrum of a sea anemone toxin which contains the 6-residues α -helical segment (14%), β -structure (36%) and unordered form (50%). The results obtained agree well with the spectrum of cmRTX-I but differ considerably from that of native RTX-I (fig.1). This difference is presumably due to an occurrence of several β -turns in the sea anemone toxins [3], model spectra of which were not available in [14].

3.2. Infrared spectroscopy

IR studies of polypeptides and proteins conducted systematically in [15-18] enabled us to obtain quantitative data concerning their secondary structure. A graphical analysis of the absorption bands in the amide I region of the RTX-I IR spectrum (in 0.1 M NaCl/D₂O solution) is shown in fig.2. The experimental contour (fig.2a (1)) has a maximum at 1636 cm⁻¹ and a shoulder at ~1650 cm⁻¹ which are indicative of the β-structure and the unordered form in the RTX-1 [15]. To obtain the peptide absorption (curve (2)) we have subtracted the absorption of the amino acid side chains (R) from the experimental contour; the R curve has been calculated in according to the known amino acid composition of RTX-I [8] and the data from [16]. The contour thus obtained (fig.2b) has been consequently decomposed into components due to the β -structure (1630 and 1660 cm⁻¹) and unordered form (1654 cm⁻¹) absorptions. Amounts of these forms (21-25% and 45-50%, respectively) have been estimated from the band integral intensities [15]. The splitting of the β -form

Table 1
Secondary structure of sea anemone toxins calculated by modified Chou-Fasman method [7]

Toxin ^a	α-Helix	β-Structure	β-Turns
ATX-I	4_9	18-27,36-43	10-13,27-30
ATX-II	4-9 (31-37) ^b	17-26,37-44	10-13,27-30
AP-A	2-9 (33-41)	16-24,38-45	10-13,25-28,29-32
AP-C	4-9 (31-37)	17-25,37-44	10-13,27-30
Mean			
content	14%	36%	19%

^a Abbreviations: ATX-I and ATX-II, toxins I and II from Anemonia sulcata; AP-A, anthopleurin A from Anthopleura xanthogrammica; AP-C, antopleurin C from Anthopleura elegantissima

b Second eventual α -helical region is shown in brackets

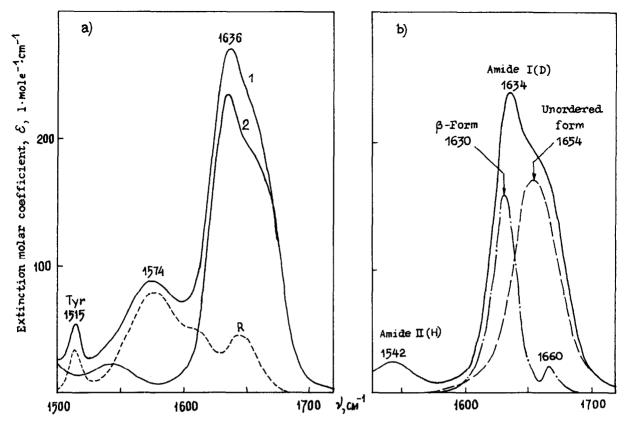


Fig. 2. Graphical analysis of IR spectrum of RTX-I in 0.1 M NaCl/D₂O solutions: (a) experimental spectrum (1) and contour of peptide absorption (2) obtained by subtraction of amino acid side chain absorption (R) from (1); (b) absorption bands of β -form (1630 and 1660 cm⁻¹) and unordered form (1654 cm⁻¹).

components which is equal to $30~\text{cm}^{-1}$ corresponds to an occurrence of 2 chains in the anti-parallel β -pleated sheet [17]. An eventual contribution of the α -helix to the peptide absorption is masked by the unordered form band [18] which makes difficult to estimate the content of the α -helical form. The obtained results on the ordered structure content may be underestimated due to distortion of the peptide unit geometry at the ends of the α - and β -regions.

3.3. Chou-Fasman calculations

The statistical Chou-Fasman method [19] was successfully applied for prediction of the secondary structure of proteins. Moreover, this method as shown in [20] gives more accurate results for smaller and thermally stable proteins to which sea anemone toxins belong [6]. We have used the modified Chou-Fasman method [7] to predict the α -helix, β -structure and β -turns in 4 sea anemone toxins with the known primary structure [1] (table 1). The calculations show that 1 α -helical, 2 β -strand regions and 2 (or 3) β -turns may be proposed for the toxins studied. A second predicted α -helical region between

the residues 31-37 (ATX-II and AP-C) or 33-41 (AP-A) may be regarded as an artifact because the analysis of spectral data on ATX-II [2] and AP-A [3] and our results do not confirm a high content of the α -helix in the sea anemone toxins.

These results indicate a similarity of the overall conformation of investigated toxins, which could explain their identical physiological activity and similar spectral properties [2–6]. The data of Chou-Fasman method and analysis of the IR spectrum of RTX-I have allowed us to suggest a model for spatial arrangement of the sea anemone toxins with antiparallel packing of 2 predicted β -strands. For example, the tertiary structure of the AP-A is shown in fig.3, for which the greatest amount of spectral data is available [3,4].

4. Discussion

In pioneer Raman spectroscopic studies of ATX-II [2], it was concluded that only the unordered form was predominant in its molecule. A study of anthopleurin A by Raman spectroscopy showed [3] that

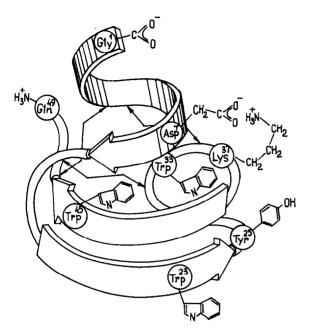


Fig. 3. Probable model of spatial arrangement of AP-A built up according to data of modified Chou-Fasman method [7]. α -Helical region is shown as spiral band and β -structure as antiparallel arrows. Disulfide bridges are shown as zig-zigged arrows [3].

its secondary structure contained α -helix (21%), β -structure (22%) and unordered form (57%). These results have been confirmed by our calculation of the AP-A secondary structure (table 1) by the modified Chou-Fasman method (16%, 34% and 50%, respectively). Analysis of the CD spectrum of AP-A suggests a probable occurrence of the β -structure and β -turns in its molecule [3] while the similarity between CD spectra of AP-A and RTX-I confirms a close relationship of the secondary structure of RTX-I and all formerly investigated sea anemone toxins.

A model of the spatial arrangement of AP-A (fig.3) proposed by us on the basis of the modified Chou-Fasman method agrees with the results of an NMR and Raman spectroscopy study of AP-A. As shown in [3,4] AP-A in solution has a roughly spherical shape with the Tyr-25 and Trp-23 residues exposed and the Trp-33 and Trp-45 residues buried for the solvent. The close located Asp-7 and Lys-37 residues form an intramolecular ionic hydrogen bond (fig.3). This part of the molecule, with its surrounding residues was assumed [5] to be 'an active site' of the AP-A and ATX-II toxins.

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References

- [1] Rathmayer, W. (1979) in: Advances in Cytopharmocology (Ceccarelli, B. and Clementi, F. eds) vol. 3, pp. 335-344, Raven Press, New York.
- [2] Prescott, B., Thomas, G. J., Beress, L., Wunderer, G. and Tu, A. T. (1976) FEBS Lett. 64, 144-147.
- [3] Ishizaki, H., McKay, R. H., Norton, T. R., Yasunobu, K. T., Lee, J. and Tu, A. T. (1979) J. Biol. Chem. 254, 9651-9656.
- [4] Norton, R. S. and Norton, T. R. (1979) J. Biol. Chem. 254, 10220-10226.
- [5] Norton, R. S., Zwick, J. and Beress, L. (1980) Eur. J. Biochem. 113, 75-83.
- [6] Nabiullin, A. A., Kozlovskaya, E. P. and Elyakov, G. B. (1980) III USSR-FRG Symp. Chem. Pept. Prot., abst. p. 31, Makhachkala, USSR.
- [7] Dufton, M. J. and Hider, R. C. (1977) J. Mol. Biol. 115, 177-193.
- [8] Zykova, T. Z., Kozlovskaya, E. P. and Elyakov, G. B. (1980) V All-Union Symp. Chem. Phys. Prot. Pept., abst. p. 70, Baku.
- [9] Grestfield, A. M., Moore, S. and Stein, W. H. (1963)J. Biol. Chem. 238, 622-627.
- [10] Argos, P., Hanei, M. and Garavito, R. M. (1978) FEBS Lett. 93, 19-24.
- [11] Chou, P. Y. and Fasman, G. D. (1977) J. Mol. Biol. 115, 135-175.
- [12] Sears, D. W. and Beychok, S. (1973) in: Physical Principles and Techniques of Protein Chemistry (Leach, S. J. ed) pt C, pp. 446-593, Academic Press, New York.
- [13] Menez, A., Langlet, G., Tamiya, N. and Fromageot, P. (1978) Biochimie 60, 505-516.
- [14] Chen, Y.-H., Yang, J. T. and Chau, K. H. (1974) Biochemistry 13, 3350-3359.
- [15] Chirgadze, Yu. N., Shestopalov, B. V. and Venyaminov, S. Yu. (1973) Biopolymers 12, 1337-1351.
- [16] Chirgadze, Yu. N., Fedorov, O. V. and Trushina, N. P. (1975) Biopolymers 14, 679-694.
- [17] Chirgadze, Yu. N. and Nevskaya, N. A. (1976) Biopolymers 15, 607-625.
- [18] Chirgadze, Yu. N. and Brazhnikov, E. V. (1974) Biopolymers 13, 1701-1712.
- [19] Chou, P. Y. and Fasman, G. D. (1974) Biochemistry 13, 222-245.
- [20] Argos, P., Schwarz, J. and Schwarz, J. (1976) Biochim. Biophys. Acta 439, 261-273.